CHEM 446 Syllabus

Course Information

Course Code	Course Name	Credits	Prerequisite(s)	Classification
CHEM 446	Electrochemistry	2	CHEM 344	Elective Course

Course Description

The course aims to give the students redox reaction, standard electrode and cell potential, Nernst equation, a different type of electrodes and electrochemical cell. Kinetics of electrical processes. Applications of electrochemistry and controlling the Corrosion reaction.

Class Scheduling:

Classes are held 2 times/week each for 50 minutes

TEXTBOOK(s)

- 1- Electrochemistry Principles, Methods and applications, C. M. A. Brett and A.M. O. Brett, 1993, New York.
- 2- Electrochemical methods: Fundamentals and Applications, Alien J. Bord, 2001, John Wiley.

Course Coordinator(s)

Prof. Aisha Ali Ganash

RELATIONSHIP TO SOs

1	2	3	4	5	6
X			X		X

CLOs

By the end of this course student will be able to:

CLO1: Identify redox reaction, cell potential, different types of electrodes and electrochemical cell. **(CH1,2,3)**. (SO1)

CLO2: Clarify all phenomena related to electrode dynamic and the corrosion reaction. (CH4,5) (SO1)

CLO3: Use the electromotive series to express the cell potential and its practical application. (CH1,3) (SO1)

CLO4: Discuss the electrochemical cell and different electrodes. (CH2,3) (SO1)

CLO5: Evaluate the rate of charge transfer and corrosion rate. **(CH4,5)** (SO1)

CLO6: Participate effectively in the group for oral presentation. (CH5) (SO6, SO4)

Contents

List of Topics	No. of Weeks
Chapter 1: Redox reaction and the Nernst equation.	3
Chapter 2: Type of electrochemical cell	3
Chapter 3: Type of Electrodes	3
Chapter 4: Electrode dynamic	4
<u>Chapter 5:</u> Corrosion	2

GRADE DISTRIBUTION: The table below is given to the students to show them their grade distribution in this course.

Week Due	Task	Score %
7	First exam	15%
12	Second exam	15%
During Semester	Assignments	20%
14	presentation	10%
16	Final exam	40%