## LAB EXPERIMENT 4

## Estimation of carbohydrate by Anthrone Method.

The anthrone reaction is the basis of a rapid and convenient method for the determination of carbohydrates, either free or present in polysaccharides.

## Principle:

Carbohydrates are dehydrated by concentrated $\mathrm{H}_{2} \mathrm{SO}_{4}$ to form furfural.Furfural condenses with anthrone to form a blue-green colored complex solution shows an absorption maximum at 620 nm , which is measured colorimetrically. note that some carbohydrates may give other colors.

The extinction depends on the compound investigated, but is constant for a particular molecule.

## Materials:

1. Anthrone reagent ( $0.2 \%$ in conc. $\mathrm{H}_{2} \mathrm{SO}_{4}$ ).
2. Glucose ( $10 \mathrm{mg} / 100 \mathrm{ml}$ ).

## Procedure:

1. Pipette out into a series of test tubes different volumes of glucose solution and make up the volume to 1 ml with water.
2. Add 4 ml of anthrone reagent to each tube.
3. mix well.
4. Cover the tubes with marbles on top to prevent loss of water by evaporation.
5. Keep the tubes in a boiling water bath for 10 minutes.
6. cool to room temperature.
7. Measure the optical density at 620 nm using a blank tube containing 1 ml water and 4 ml reagent.
8. Draw the standard cure and determine the concentration of unknown glucose solution

Name:
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Experiment 4:

Results Sheet


