

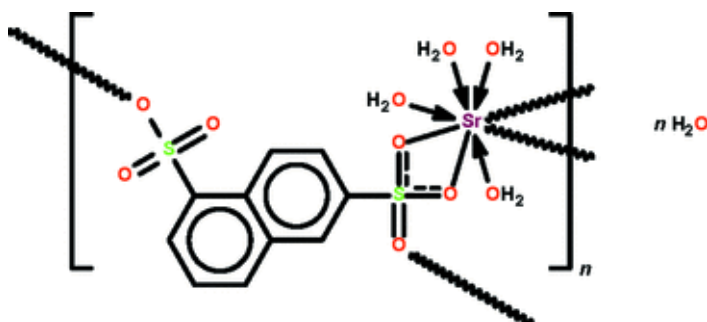
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metal-organic compounds

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Poly[[tetraaqua(μ_3 -naphthalene-1,6-disulfonato- $^4O^1:O^6,O^{6'}:O^{6''}$)strontium(II)] monohydrate]

S. Gao and S. W. Ng

Abstract: In the crystal structure of the polymeric title compound, $\{[\text{Sr}(\text{C}_{10}\text{H}_6\text{O}_6\text{S}_2)(\text{H}_2\text{O})_4]\cdot\text{H}_2\text{O}\}_n$, the naphthalene-1,6-disulfonate dianion uses one $-\text{SO}_3$ unit to O,O' -chelate to an Sr^{II} cation and its third O atom to bind to another Sr^{II} cation. The other $-\text{SO}_3$ unit binds to yet another Sr^{II} atom. The four coordinated water molecules are monodentate but one is disordered over two positions in a 1:1 ratio. The μ_3 -bonding mode of the dianion generates a polymeric three-dimensional network; the network is consolidated by $\text{O}-\text{H}\cdots\text{O}$ hydrogen bonds. The Sr^{II} cation exists in an undefined eight-coordinate environment.